

5-Benzylidene-1-phenyl-2'',3''-dihydro-
piperidine-3-spiro-2'-pyrrolizidine-3'-
spiro-3''-1H-indole-4,2''-dioneV. Vinodh Kumar,^a D. Gayathri,^a
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Key indicators

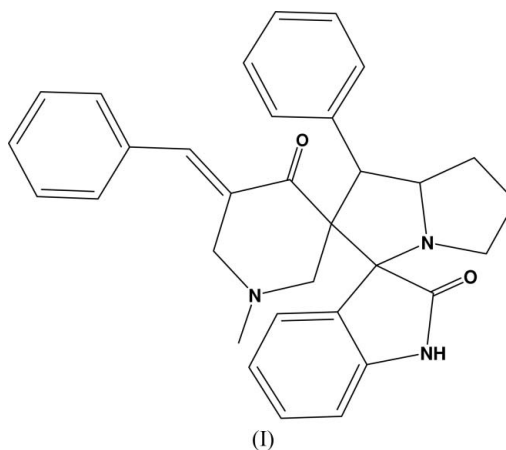
Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.052
 wR factor = 0.142
Data-to-parameter ratio = 18.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the pyrrolizidine unit of the title compound, $\text{C}_{32}\text{H}_{31}\text{N}_3\text{O}_2$, one pyrrolidine ring adopts an envelope conformation and the other adopts a twist conformation. The molecular conformation is stabilized by weak $\text{C}-\text{H}\cdots\text{O}$ intramolecular interactions and the crystal packing is stabilized by $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

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Comment

Pyrrolidines are common structural motifs in drugs and drug candidates, showing antimicrobial and antifungal activity (Amal Raj *et al.*, 2003). Oxindole derivatives are found to be potent aldose reductase inhibitors (ARIs), which help to treat and prevent diabetic complications arising from elevated levels of sorbitol (Rajeswaran *et al.*, 1999).



The geometric parameters of the title compound are comparable with the literature values (Allen *et al.*, 1987), except at the spiro junctions. The sum of the bond angles around N1 [340.5°] and N3 [334.2°] indicates sp^3 hybridization of these N atoms. The two phenyl rings (C18–C23 and C27–C32) are almost perpendicular to each other, making a dihedral angle of 84.0 (1)°.

One pyrrolidine ring (N1/C1–C4) adopts an envelope conformation, with atom C4 deviating by 0.662 (2) Å from the mean plane through the remaining atoms. The other pyrrolidine ring (N1/C4–C7) adopts a twist conformation with a local twofold axis passing through the N1–C4 bond and atom C6. The puckering parameters (Cremer & Pople, 1975) and the smallest displacement asymmetry parameters (Nardelli, 1983) for the pyrrolidine rings (N1/C1–C4 and N1/C4–C7) are $q_2 = 0.440$ (2)/0.424 (2) Å, $\varphi = 330.8$ (2)/20.8 (2) and $\Delta_s(\text{C}_4) = 10.4$ (1) and $\Delta_2(\text{C}_6) = 2.5$ (2), respectively.

The molecular conformation is stabilized by weak C—H···O intramolecular interactions. The crystal packing is stabilized by N—H···O and C—H···O hydrogen bonds. Atom N2 acts as a donor to O1 at (1 - x, 1 - y, 1 - z), generating a centrosymmetric dimer of the type $R_2^2(8)$.

Experimental

Isatin (1 mmol), racemic proline (1 mmol), 3,5-dibenzylidene)-1-methylpiperidin-4-one (1 mmol) and 1 g of bentonite clay were mixed thoroughly in a boiling tube which was loosely closed and immersed in a silica-gel bath in a beaker and irradiated in a domestic microwave oven for about 10 min at a maximum power level (600 W) in two 5 min periods with intermittent cooling. The reaction mixture was then cooled to room temperature and tested for the completion of reaction. The product was isolated by column chromatography using *n*-hexane–ethyl acetate in a 5:1 ratio, and was recrystallized (m.p. 441 K) from a hexane–ethyl acetate (4:1) mixture.

Crystal data

$C_{32}H_{31}N_3O_2$	$V = 2643.3(2) \text{ \AA}^3$
$M_r = 489.60$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 8.7120(4) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 20.5711(10) \text{ \AA}$	$T = 293(2) \text{ K}$
$c = 14.9467(7) \text{ \AA}$	$0.25 \times 0.24 \times 0.22 \text{ mm}$
$\beta = 99.322(1)^\circ$	

Data collection

Bruker SMART CCD area-detector diffractometer	6227 independent reflections
Absorption correction: none	4738 reflections with $I > 2\sigma(I)$
30223 measured reflections	$R_{int} = 0.025$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.052$	334 parameters
$wR(F^2) = 0.142$	H-atom parameters constrained
$S = 1.01$	$\Delta\rho_{max} = 0.26 \text{ e \AA}^{-3}$
6227 reflections	$\Delta\rho_{min} = -0.14 \text{ e \AA}^{-3}$

Table 1

Selected bond angles ($^\circ$).

C4—N1—C1	108.8(1)	C26—N3—C25	112.5(1)
C4—N1—C7	107.6(1)	C26—N3—C24	110.7(1)
C1—N1—C7	124.1(1)	C25—N3—C24	111.0(1)

Table 2

Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3···O2	0.98	2.32	2.782(2)	108
C17—H17···O2	0.93	2.38	2.759(2)	104
C25—H25B···O1	0.97	2.38	2.926(2)	116
N2—H2···O1 ⁱ	0.86	2.05	2.874(2)	161
C19—H19···O2 ⁱⁱ	0.93	2.47	3.226(2)	139

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x, -y + 1, -z$.

All H atoms were refined using a riding model, with C—H = 0.93 \AA and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic, C—H = 0.98 \AA and $U_{iso}(H) = 1.2U_{eq}(C)$ for CH, C—H = 0.97 \AA and $U_{iso}(H) = 1.2U_{eq}(C)$ for CH_2 ,

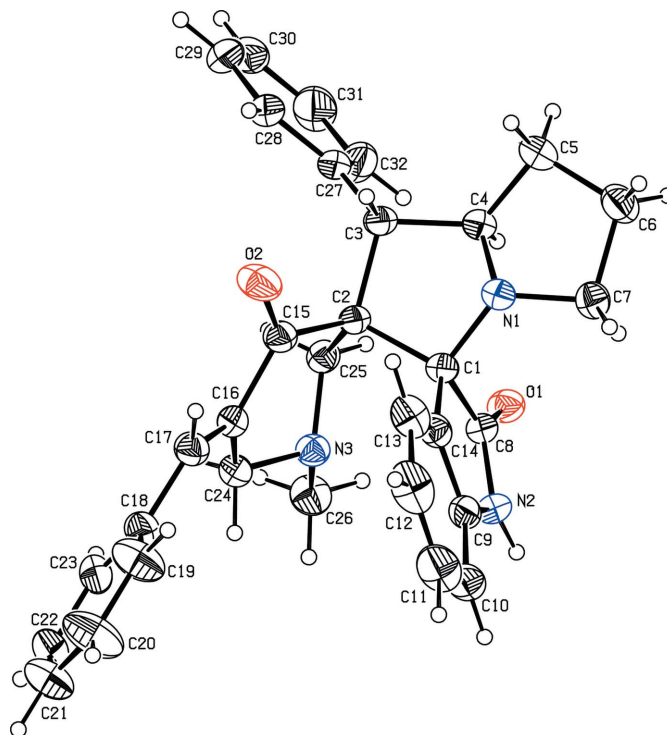


Figure 1

The molecular structure of the title compound, showing 30% probability displacement ellipsoids.

C—H = 0.96 \AA and $U_{iso}(H) = 1.5U_{eq}(C)$ for CH_3 , and N—H = 0.86 \AA and $U_{iso}(H) = 1.2U_{eq}(N)$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

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